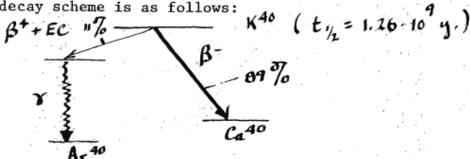
(1) Consider the problem of determining the age of a sample of the earth (a rock) from the relative abundance of K⁴⁰ and Ar⁴⁰ in that rock.

We may safely assume that while the material is molten or gaseous, the Ar⁴⁰ escapes. Thus Ar⁴⁰ slowly builds up from zero abundance starting at the time the rock material last solidified. The relevant decay scheme is as follows:



If one studies a large sample of rocks, one finds that the molar abundance ratio $\frac{A}{K}$ of Ar^{40} to K^{40} is always less than a limiting value:

$$\left(\frac{A}{K}\right)_{max} = 1.23$$

Using this fact and the decay scheme above, first derive a general formula which expresses the age of the rock t in terms of $\frac{A}{K}$ and the partial decay rates to Ar^{40} , Ca^{40} ; then compute the maximum rock age t_{max} (= t_{EARTH}).

