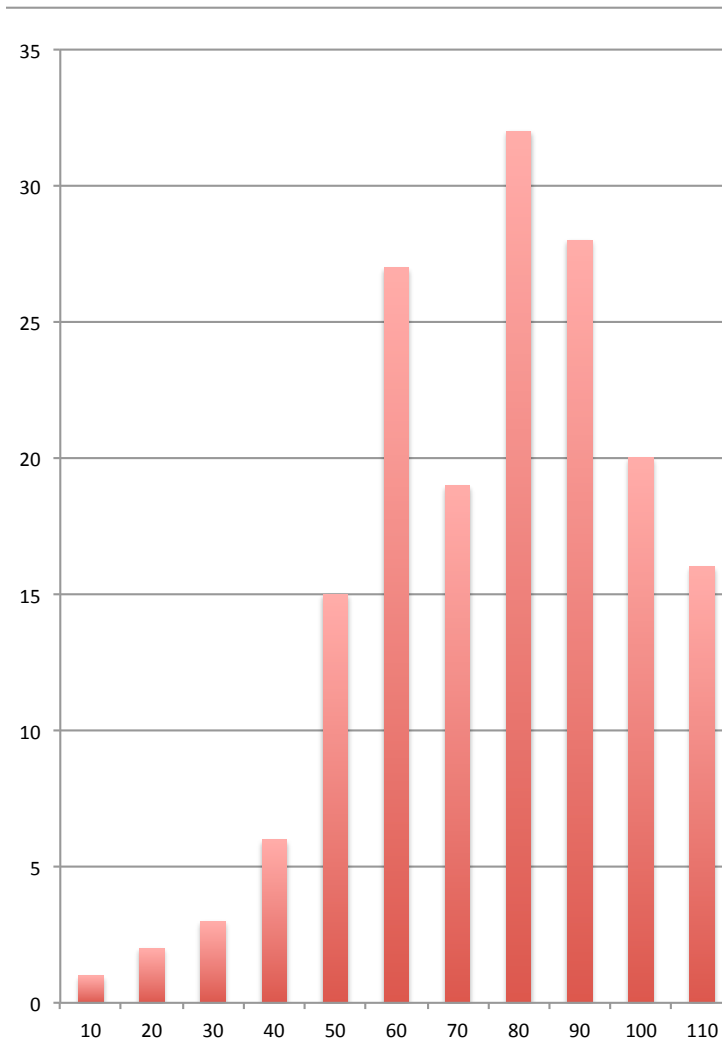


Lecture 10

PHYC 161 Fall 2016

Exam I

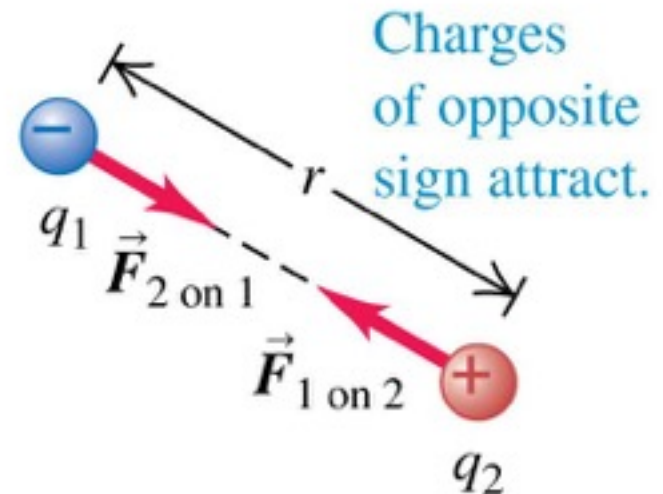
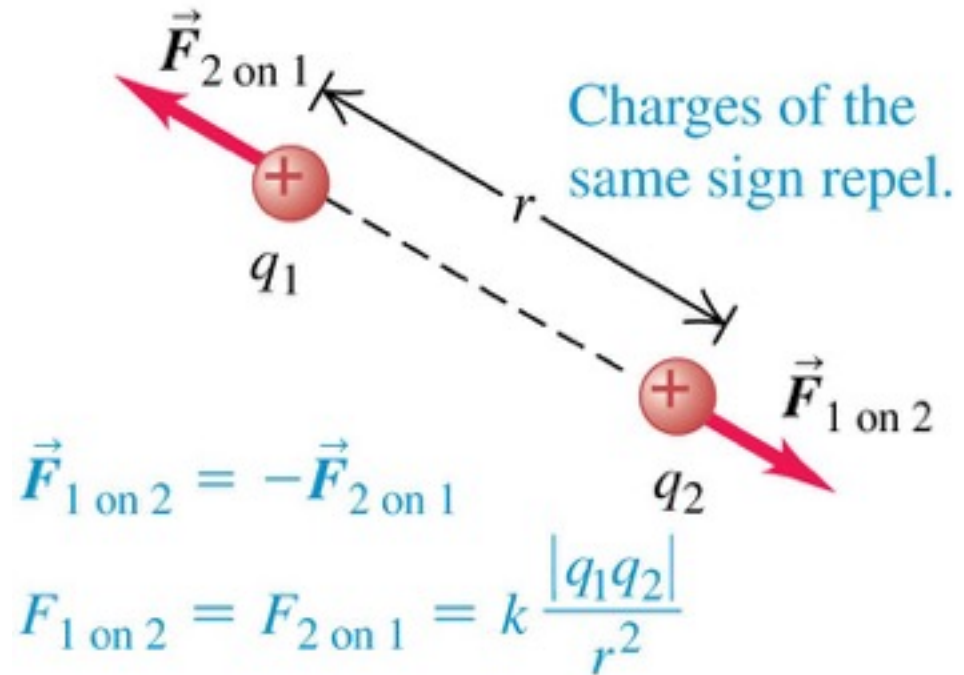


Mean = 76

Coulomb's Law

- **Coulomb's Law:** The magnitude of the electric force between two point charges is directly proportional to the product of their charges and inversely proportional to the square of the distance between them.

$$F = k \frac{|q_1 q_2|}{r^2}$$



The electric field of a point charge

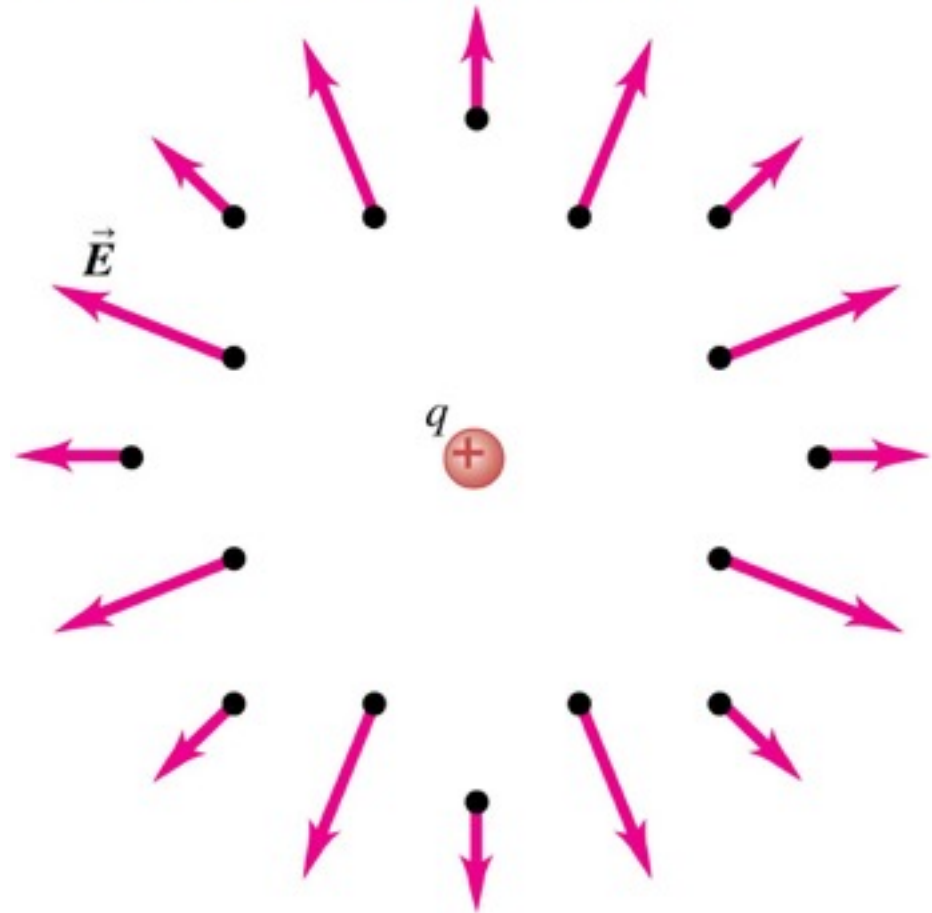
- Using a unit vector that points away from the origin, we can write a vector equation that gives both the magnitude and the direction of the electric field:

The diagram shows the equation for the electric field \vec{E} due to a point charge q . The equation is
$$\vec{E} = \frac{1}{4\pi\epsilon_0} \frac{q}{r^2} \hat{r}$$
 The terms in the equation are labeled with blue text and arrows:

- Electric field due to a point charge** points to \vec{E} .
- Value of point charge** points to q .
- Electric constant** points to ϵ_0 .
- Distance from point charge to where field is measured** points to r^2 .
- Unit vector from point charge toward where field is measured** points to \hat{r} .

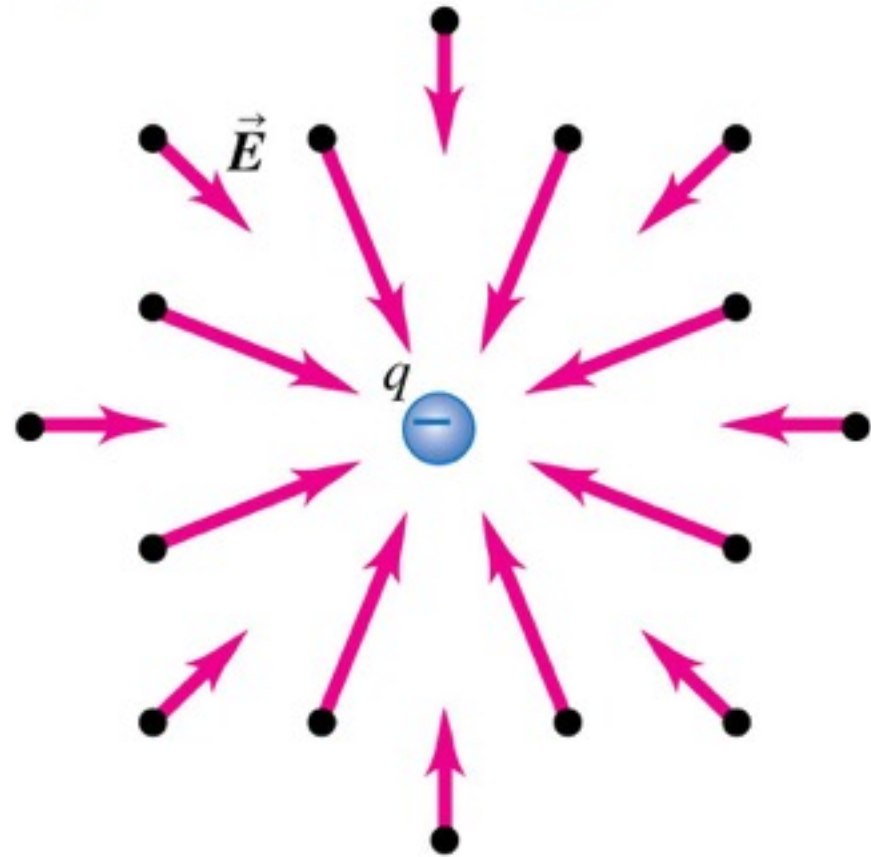
Electric field of a point charge

- A point charge q produces an electric field at *all* points in space.
- The field strength decreases with increasing distance.
- The field produced by a positive point charge points *away from* the charge.



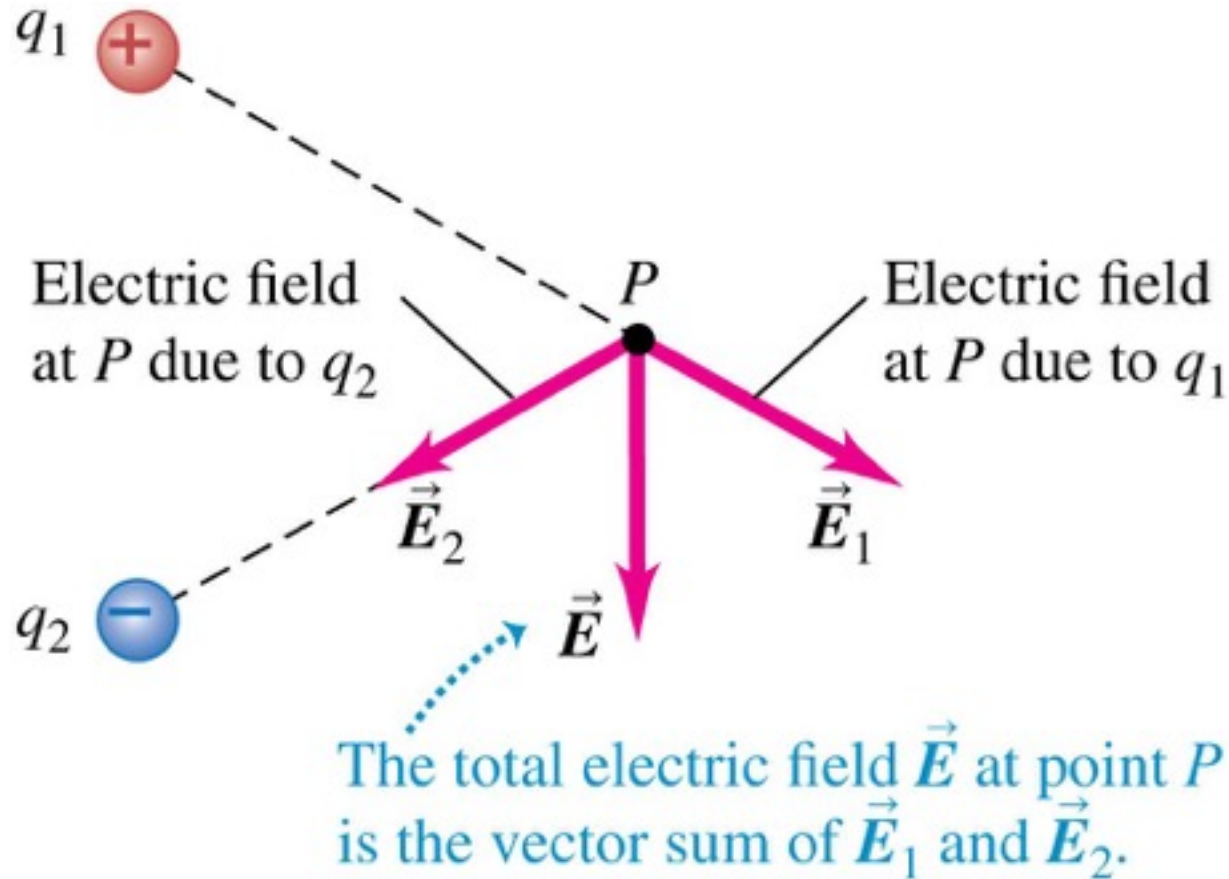
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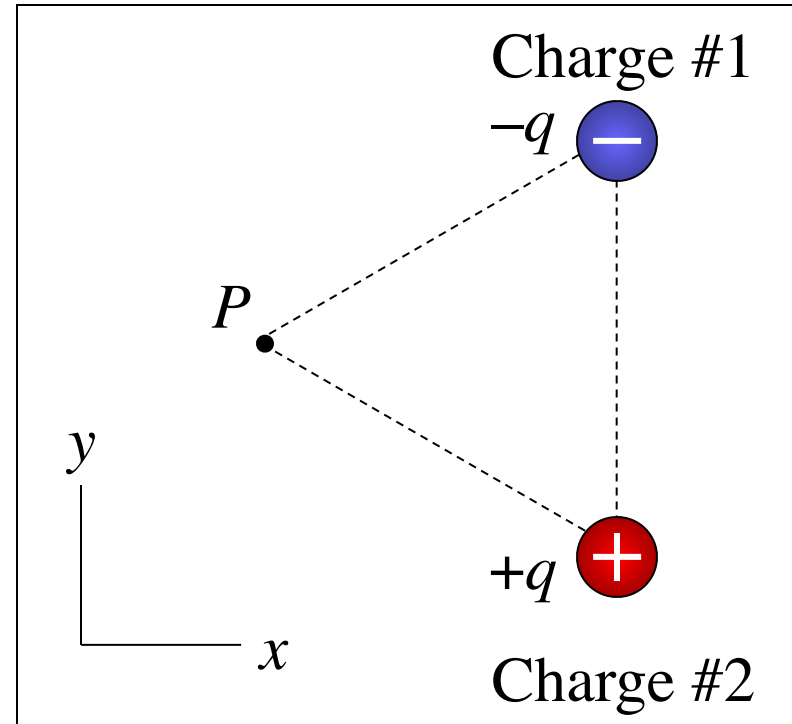
Superposition of electric fields

- The total electric field at a point is the vector sum of the fields due to all the charges present.



Q21.6

Two point charges and a point P lie at the vertices of an equilateral triangle as shown. Both point charges have the same magnitude q but opposite signs. There is nothing at point P . The net electric field that charges #1 and #2 produce at point P is in



A. the $+x$ -direction.

B. the $-x$ -direction.

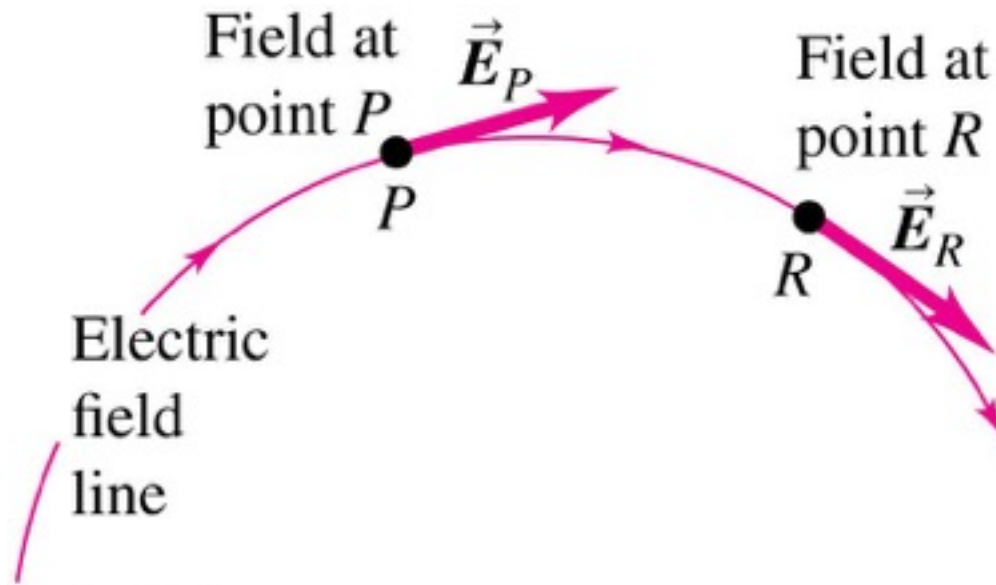
C. the $+y$ -direction.

D. the $-y$ -direction.

E. none of the above.

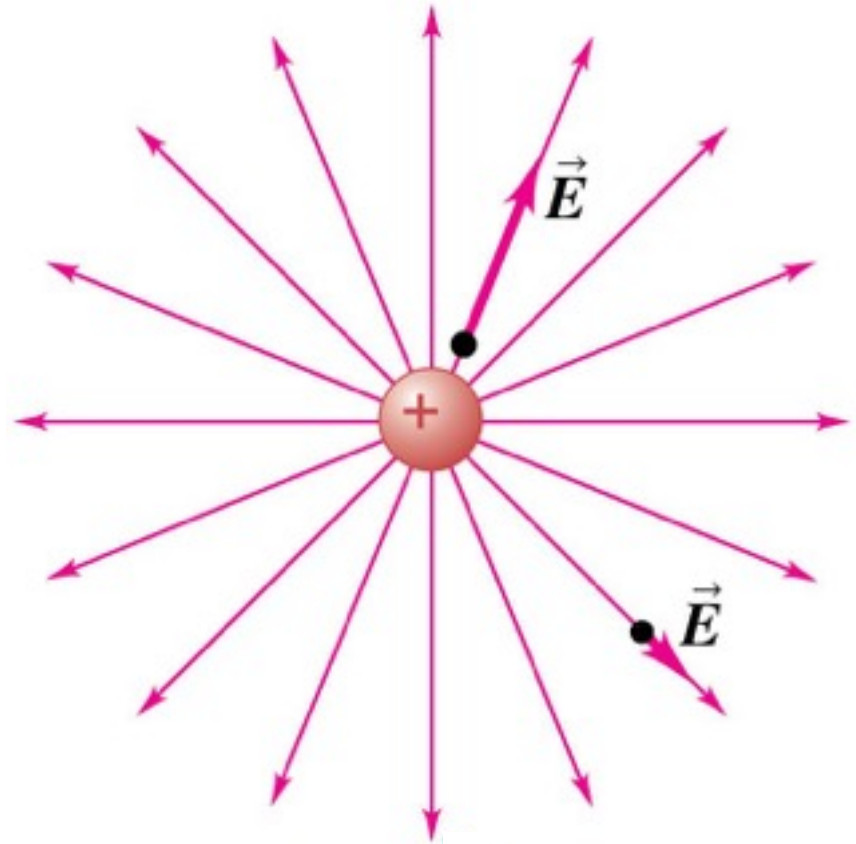
Electric field lines

- An **electric field line** is an imaginary line or curve whose tangent at any point is the direction of the electric field vector at that point.



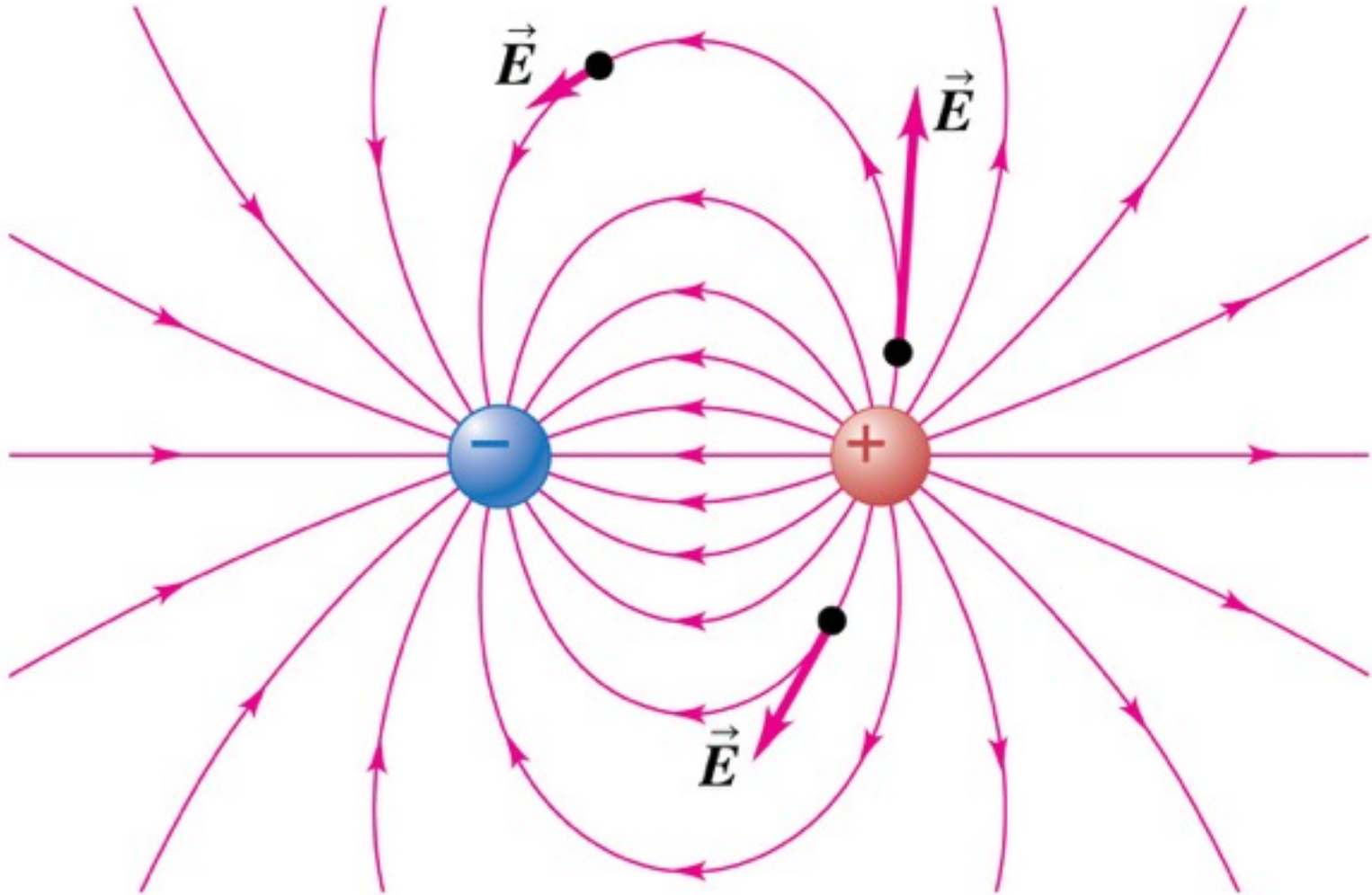
Electric field lines of a point charge

- Electric field lines show the *direction* of the electric field at each point.
- The spacing of field lines gives a general idea of the *magnitude* of the electric field at each point.



Electric field lines of a dipole

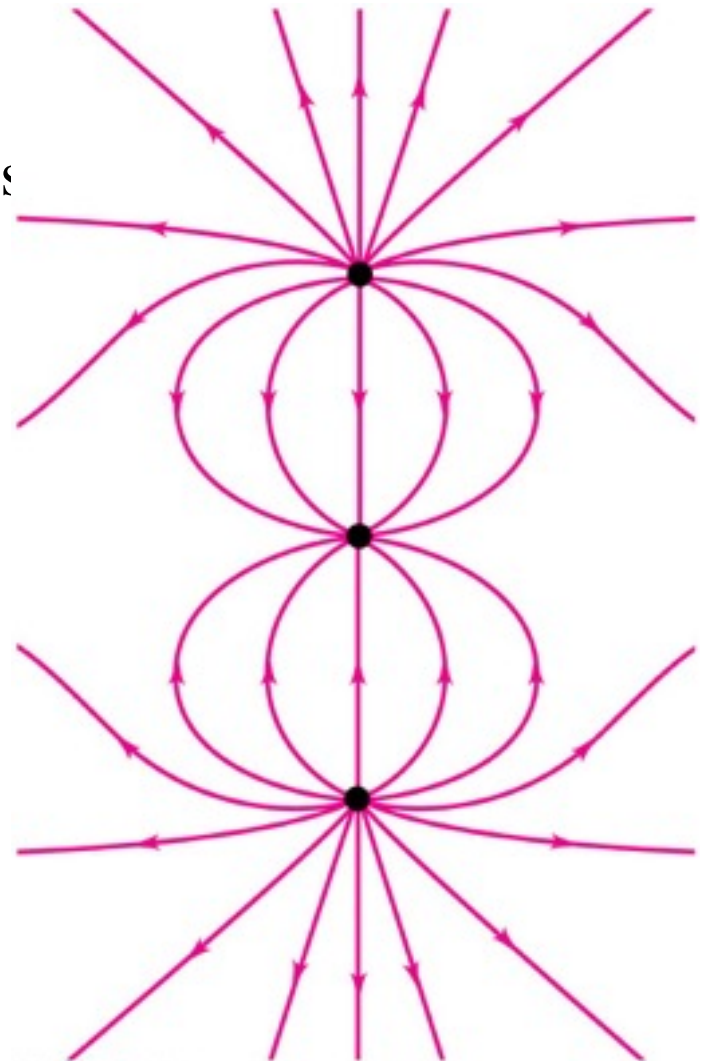
- Field lines point away from + charges and toward – charges.



Q21.9

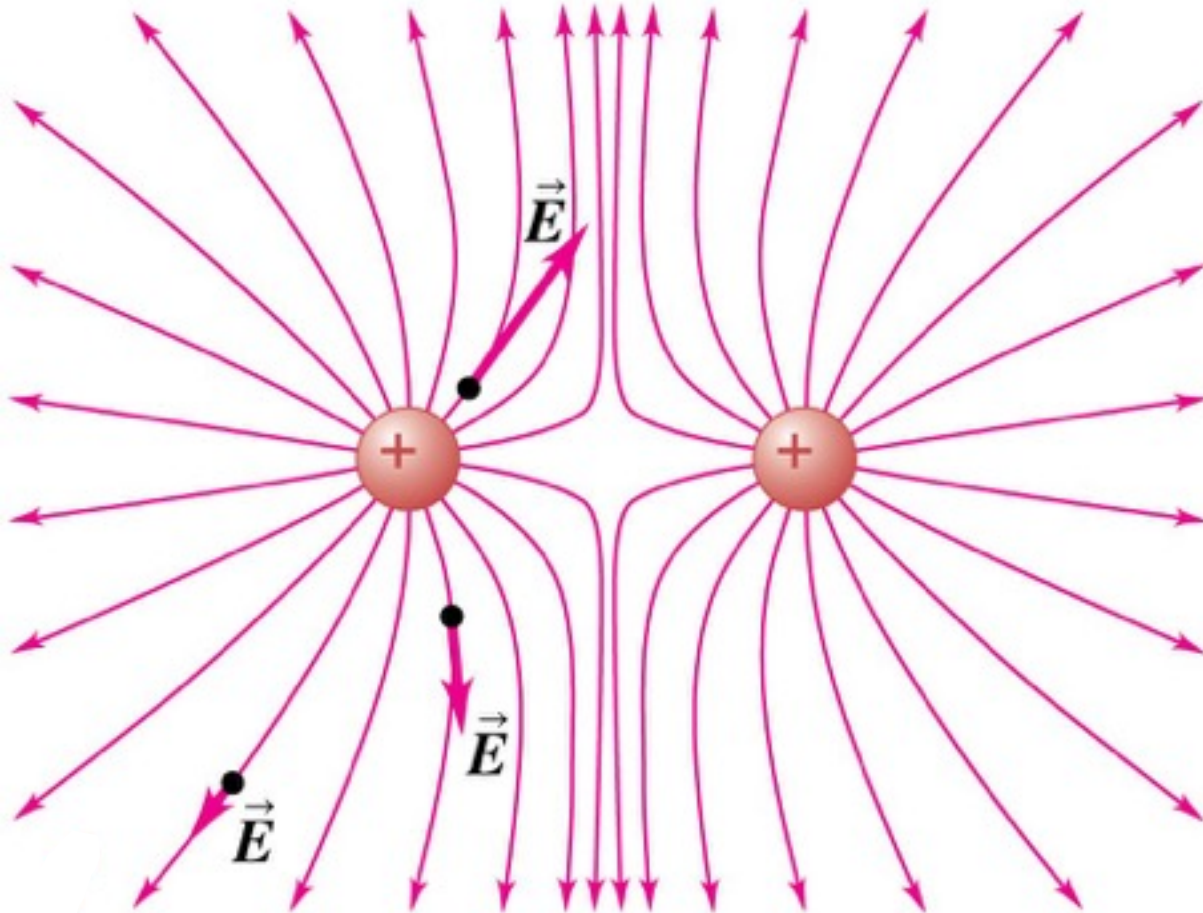
The illustration shows the electric field lines due to three point charges (shown by the black dots). The electric field is strongest

- A. where adjacent field lines are closest together.
- B. where adjacent field lines are farthest apart.
- C. where adjacent field lines are parallel.
- D. where the field lines are most strongly curved.
- E. at none of the above locations.



Electric field lines of two equal positive charges

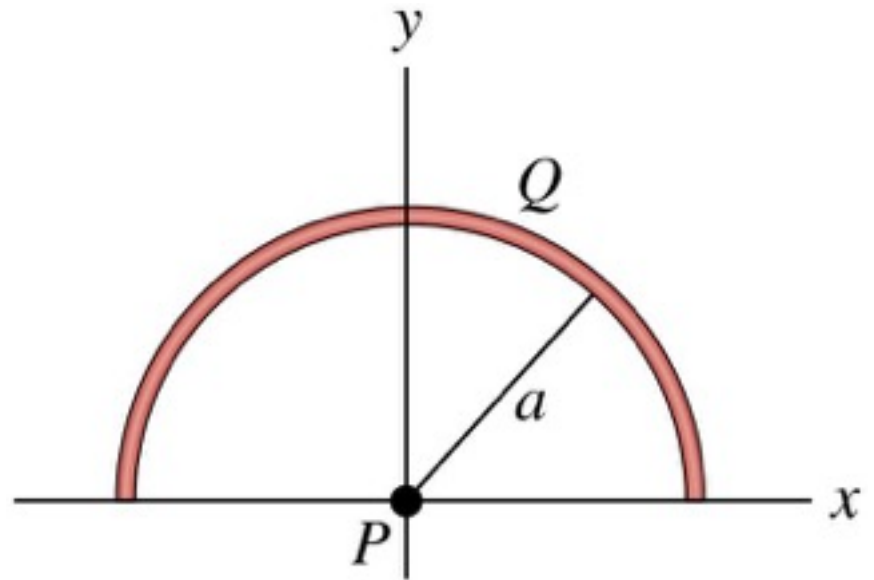
- At any point, the electric field has a unique direction, so *field lines never intersect*.



Q21.10

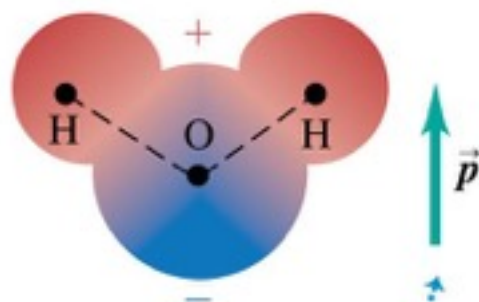
Positive charge is uniformly distributed around a semicircle. The electric field that this charge produces at the center of curvature P is in

- A. the $+x$ -direction.
- B. the $-x$ -direction.
- C. the $+y$ -direction.
- D. the $-y$ -direction.
- E. none of the above.



The water molecule is an electric dipole

- The water molecule as a whole is electrically neutral, but the chemical bonds within the molecule cause a displacement of charge.
- The result is a net negative charge on the oxygen end of the molecule and a net positive charge on the hydrogen end, forming an electric dipole.



The electric dipole moment \vec{p} is directed from the negative end to the positive end of the molecule.

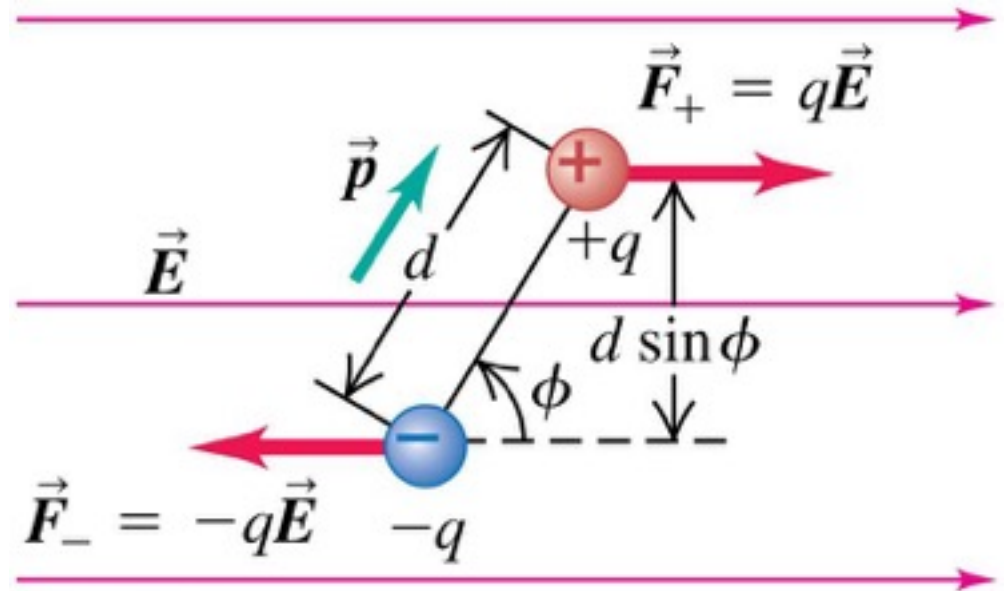
The water molecule is an electric dipole

- When dissolved in water, salt dissociates into a positive sodium ion and a negative chlorine ion, which tend to be attracted to the negative and positive ends of water molecules.
- This holds the ions in solution.
- If water molecules were not electric dipoles, water would be a poor solvent, and almost all of the chemistry that occurs in aqueous solutions would be impossible!



Force and torque on a dipole

- When a dipole is placed in a uniform electric field, the net *force* is always zero, but there can be a net *torque* on the dipole.



$$\vec{F}_g = m_0 \vec{g}$$